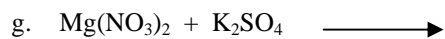
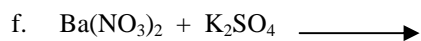
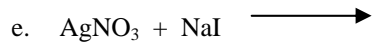
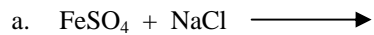


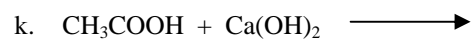
CHEMISTRY 123-01

Exercise #2

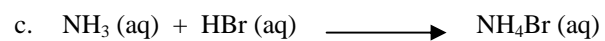
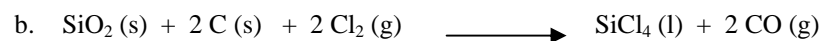
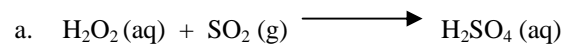
September 21, 2007

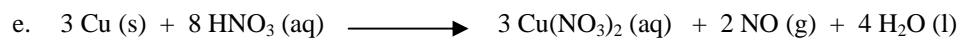
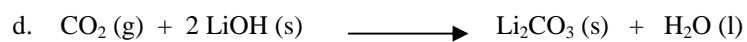
1. Write the complete and net ionic equations for each of the following reactions (in water). If no reaction occurs, write NR after the arrow.





2. Find which of the equations below represent redox reactions. Indicate the reducing and oxidizing agent.





3. For each of the following displacement reactions, predict whether the process will occur from left to right or from right to left.

